

UNCHING INDUSTRY CO., LTD.

109/20 Moo 9, Setthakit 1 Rd., T. Suanluang, A.Krathumbaen, Samutsakhon 74110.

Tel: 02 - 810 - 1345 Fax: 02 - 810 - 1346

SAFETY DATA SHEET

COPPER SULPHATE PENTAHYDRATE

1. Identification of the substance/preparation and of the company/undertaking

Product name: COPPER SULPHATE PENTAHYDRATE

Manufacturer/supplier identification

Company: UNCHING INDUSTRY CO., TLD

109/20 Moo 9, Setthakit 1 Rd., T. Suanluang, A.Krathumbaen,

Samutsakhon 74110

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2. Hazards identification

GHS classification of the substance/mixture

Health hazards

Acute Toxicity (Oral) Category 4

Serious eye damage/eye irritation Category 1

Environmental hazards

Acute aquatic toxicity Category 1

Chronic aquatic toxicity Category 1

Pictogram (s)



Signal word:

Danger

Hazard statements:

H302 Harmful if swallowed.

H318 Causes serious eye damage.

H410 Very toxic to aquatic life with long lasting effects.

Precautionary statements

P264 Wash skin thoroughly after handling.

P270 Do not eat, drink or smoke when using this product.

P273 Avoid release to the environment.

P280 Wear eye protection/ face protection.

P301 + P312 IF SWALLOWED: Call a POISON CENTER/ doctor if you feel unwell.

P305 + P351 + P338 IF IN EYES: Rinse cautiously with water for several minutes.

Remove contact lenses, if present and easy to do. Continue rinsing.

3. Composition/information on ingredients

	Product Name	CAS Numbers	EC Numbers	Index-No.	%Copper
COPPER	SULPHATE PENTAHYDRATE	7758-99-8	231-847-6	029-023-00-4	25

Formula CuSO₄.5H₂O

Molecular Weight 249.68 AMU

Synonyms Blue copper AS * Copper(II) sulfate pentahydrate * Copper(2+) sulfate pentahydrate * Cupric sulfate pentahydrate * Kupfersulfat-pentahydrat (German) * Kupfervitriol (German) * Salzburg vitriol * Sulfuric acid, copper(2+) salt, pentahydrate * Sulfuric acid copper(2+) salt (1:1), pentahydrate (8CI,9CI) * Vencedor

4. First aid measures

After inhalation: Fresh air. If inhaled, remove to fresh air. If not breathing give artificial respiration. If breathing is difficult, give oxygen.

After skin contact: In case of skin contact, flush with copious amounts of water for at least 15 minutes. Remove contaminated clothing and shoes. Call a physician.

After eye contact: In case of contact with eyes, flush with copious amounts of water for at least 15 minutes. Assure adequate flushing by separating the eyelids with fingers. Call a physician.

After ingestion: If swallowed, wash out mouth with water provided person is conscious. Call a physician.

5. Fire-fighting measures

Extinguishing media

Suitable: Water spray. Carbon dioxide, dry chemical powder, or appropriate foam.

Special risks

and eyes.

Specific Hazard(s): Emits toxic fumes under fire conditions.

Special risks special protective equipment for firefighters.

Wear self-contained breathing apparatus and protective clothing to prevent contact with skin

6. Accidental release measures

Personal precaution procedures to be followed in case of leak or spill.

Evacuate area.

Procedure (s) of personal precaution (s).

Wear self-contained breathing apparatus, rubber boots, and heavy rubber gloves.

Methods for cleaning up.

Sweep up, place in a bag and hold for waste disposal. Ventilate area and wash spill site after material pickup is complete.

7. Handling and storage

Handling

Directions for Safe Handling

Do not breathe dust. Avoid contact with eyes, skin, and clothing. Avoid prolonged or repeated exposure.

Storage

Conditions of Storage: Keep tightly closed. Store under argon. Store in a cool dry place.

Special requirements

Air sensitive. Hygroscopic. Handle and store under inert gas.

8. Exposure controls/personal protection

Engineering controls

Safety shower and eye bath. Mechanical exhaust required.

General hygiene measures

Wash thoroughly after handling.

Exposure limits - United kingdom

Source	1 ype	value
OEL	LTEL	$0.2 \text{ mg (Cu)/m}^3 1 \text{ mg (Cu)/m}^3$

Remarks: fumes

Personal protective equipment

Respiratory Protection: Government approved respirator.

Hand Protection: Compatible chemical-resistant gloves.

Eye Protection: Chemical safety goggles.

9. Physical and chemical properties

Sical al	iu chemicai properties			
	Appearance	Color: Blue Form: Crystals		
Property		Value	At Temperature or Pressure	
	pH	3.5 - 4.5	50 g/l	
	BP/BP Range	N/A		
	MP/MP Range	110 °C	1001	
	Flash Point	N/A		
	Flammability	N/A		
	Autoignition Temp	N/A		
	Oxidizing Properties	N/A		
	Explosive Properties	N/A		
	Explosion Limits	N/A		
	SG/Density	2.284 g/cm ³		
	Partition Coefficient	N/A		
	Viscosity	N/A		
	Vapor Density	N/A		
	Saturated Vapor Conc.	N/A		
	Evaporation Rate	N/A		
	Bulk Density	N/A		
	Decomposition Temp.	N/A		
	Solvent Content	N/A		
	Water Content	N/A		
	Surface Tension	N/A		
	Conductivity	N/A		
	Miscellaneous Data	N/A		
	Solubility	N/A		

10. Stability and reactivity

Stability

Stable: Stable.

Conditions to Avoid: Air sensitive. Hygroscopic.

Materials to Avoid: Finely powdered metals Anhydrous copper (II) sulfate reacts violently with: hydroxylamine, magnesium.

Hazardous decomposition products

Hazardous Decomposition Products: Copper oxide, Sulfur oxides.

Hazardous polymerization

Hazardous Polymerization: Will not occur.

11. Toxicological information

Acute toxicity

LDLO Oral Human 1088 mg/kg

Remarks: Behavioral: Coma.

Liver:Jaundice (or hyperbilirubinemia) hepatocellular.

Blood:Other hemolysis with or withot anemia.

Oral 300 mg/kg LD_{50} Rat Skin > 2000 mg/kg LD_{50} Rat LD_{50} Intraperitoneal Rat 18700 ug/kg LD_{50} Intraperitoneal Mouse 33 mg/kg LD_{50} Intraperitoneal Mammal 7500 ug/kg

Remarks: Behavioral:Somnolence (general depressed activity).

Gastrointestinal: Changes in structure or function of salivary glands.

Nausea or vomiting.

Sensitization

Skin: Prolonged or repeated exposure may cause allergic reactions in certain sensitive individuals.

Signs and symptoms of exposure

Chronic copper poisoning is typified by hepatic cirrhosis, brain damage and demyelination, kidney defects, and copper deposition in the cornea as exemplified by humans with Wilson's disease. It has also been reported that copper poisoning has lead to hemolytic anemia and accelerates arteriosclerosis. To the best of our knowledge, the chemical, physical, and toxicological properties have not been thoroughly investigated.

Route of exposure

Skin Contact: Causes skin irritation.

Skin Absorption: May be harmful if absorbed through the skin.

Eye Contact: Causes eye irritation.

Inhalation: May be harmful if inhaled. Material may be irritating to mucous membranes and

upper respiratory tract.

Ingestion: Harmful if swallowed.

12. Ecological information

Ecotoxicological effects

Test Type: LC₅₀ Fish

Time: 96 h

Value: 1 - 2.5 mg/l

Test Type: EC₅₀ Daphnia

Species: Daphnia magna

Time: 48 h

Value: 0.024 mg/l

13. Disposal considerations

Substance Disposal: Contact a licensed professional waste disposal service to dispose of this material. Dissolve or mix the material with a combustible solvent and burn in a chemical incinerator equipped with an afterburner and scrubber. Observe all federal, state, and local environmental regulations.

14. Transport information

RID/ADR

UN#: 3077

Class: 9

PG: III

Proper Shipping Name: Environmentally hazardous substance, solid, n.o.s

IMDG

UN#: 3077

Class: 9

PG: III

Proper Shipping Name: Environmentally hazardous substance, solid, n.o.s

Marine Pollutant: No

Severe Marine Pollutant: No

Technical Name: Required

IATA

UN#: 3077

Class: 9

PG: III

Proper Shipping Name: Environmentally hazardous substance, solid, n.o.s

Inhalation Packing Group I: No

Technical Name: Required

15. Regulatory information

Labelling according to EC Directives

Index Number: 029-023-00-4

Indication of danger: Xn N

Harmful. Dangerous for the environment.

R-Phrases: 22 36/38 50/53

Harmful if swallowed. Irritating to eyes and skin. Very toxic to aquatic organisms, may cause long-term adverse effects in the aquatic environment.

S- Phrases: 22 60 61

Do not breathe dust. This material and its container must be disposed of as hazardous waste. Avoid release to the environment. Refer to special instructions/safety data sheets.

Country specific information

Germany

WGK: 2

Switzerland

Swiss poison class: 3

16. Other information

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